



Structure and electrical properties of MnO₂-doped Sr_{2-x}Ca_xNaNb₅O₁₅ lead-free piezoelectric ceramics

Xiujun Fan^{a,*}, Yue Wang^a, Yijian Jiang^b

^a College of Applied Sciences, Beijing University of Technology, Beijing 100124, China

^b Institute of Laser Engineering, Beijing University of Technology, Beijing 100124, China

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ABSTRACT

Lead-free piezoelectric ceramics Sr_{2-x}Ca_xNaNb₅O₁₅ + y wt% MnO₂ have been prepared by the conventional solid state reaction method. Our results reveal that Ca²⁺ and Mn ions have entered into the Sr₂NaNb₅O₁₅ lattices to form a solid solution with tungsten–bronze structure. The substitution of Ca²⁺ induces a decrease in piezoelectric coefficient *d*₃₃, electromechanical coupling factors *k*_p and *k*_t, while, the addition of Mn ions decreases the sintering temperature and effectively promotes the densification of the ceramics. The effect of substitution of Ca²⁺ and Mn ions on the structure, electrical properties and diffused phase changing were investigated systematically. For the ceramic with *x* = 0.05 and *y* = 0.5, the piezoelectric, dielectric and ferroelectric properties become optimum, giving a piezoelectric coefficient *d*₃₃ = 190 pC/N, electromechanical coupling factors *k*_p = 13.4% and *k*_t = 36.5%, ε_r = 2123, loss tangent tan δ = 0.038, remanent polarization *P*_r = 4.76 μC/cm², coercive field *E*_c = 12.68 kV/cm, and Curie temperature *T*_c = 260 °C.

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1. Introduction

Lead–zirconium–titanium (PZT) based piezoceramics have excellent piezoelectric properties and hence have been widely used in the manufacture of actuators, sensors, transducers and other electromechanical devices. However, the use of the lead-based ceramics has caused serious environmental problem because of the toxicity of lead oxide and its high vapor pressure during processing [1]. Therefore, it is urgent to develop environmental benign alternative materials [2]. In recent years, a number of extensive studies on lead-free piezoelectric ceramics such as Bi_{0.5}Na_{0.5}TiO₃-based ceramics [3,4], Bi-layered structure materials [5], tungsten–bronze-type materials [6,7], BaTiO₃-based ceramics [8], and alkaline niobate-based materials [9,10] have been carried out. One of the lead-free ferroelectric oxides, mixed niobates with tungsten–bronze (TB) structure have received considerable attention owing to their potential applications as ferroelectric, piezoelectric, nonlinear optic and pyroelectric materials because of their large spontaneous polarization, high piezoelectric, electric–optic and pyroelectric coefficient [11,12]. Among these tungsten–bronze niobates, Sr_{2-x}Ca_xNaNb₅O₁₅ (abbreviated as SCNN), with high Curie temperature *T*_c (about 300 °C), large crystalline anisotropy and high piezoelectric properties along the

c-axis of the crystal structure have received increasing interests. The single crystals of SCNN have already been synthesized by Neurgaonkar et al., and found they have relatively large and nearly equal longitudinal (*d*₃₃ = 270 pC/N) and shear (*d*₁₅ = 248 pC/N) piezoelectric coefficients [13]. However, it is difficult to obtain sufficiently large single crystal of SCNN. At the same time, polycrystalline ceramics having many advantages over single crystals, such as low cost, ease of fabrication, large size, and isotropic properties. However, there are still many difficulties encountered during producing tungsten–bronze ferroelectric ceramics (e.g., Ba_xSr_{1-x}Nb₁₀O₃₀ [14], KSr₂NaNb₅O₁₅ [15], and (Pb_{1-x}Ba_x)Nb₂O₆ [16], such as abnormal grain growth, crack generation and pore formation. In order to produce pore-free and high-density SCNN ceramics, several strategies have been used, including spark plasma sintering (SPS) [17,18].

MnO₂ is one of the most interesting and useful additives, because Mn ions have multivalence: Mn²⁺, Mn³⁺, and Mn⁴⁺. When Mn ions are used to substitute the metal cations in perovskite-type ABO₃ ferroelectric materials, they generally substitute cations with similar ionic radii in the A or B site [9]. In the lead–zirconium–titanium (PZT) [19,20] and barium–titanium (BT) [21] based piezoelectric ceramics, MnO₂ additive is usually recognized as an acceptor or “hard” doping element. On the other hand, Mn doping can cause “soft” and “hard” effects simultaneously in PZT ceramics, when the Mn doping ratio is at an optimum content [22]. However, the effects of Mn doping in the tungsten–bronze structure compounds have not been studied extensively as much as that in PZT-based or BT-based ferroelectric ceramics. In the present work,

* Corresponding author. Tel.: +86 10 67392201; fax: +86 10 67392030.
E-mail address: fxiujun@gmail.com (X. Fan).

MnO₂ doped Sr_{2-x}Ca_xNaNb₅O₁₅ ceramics were synthesized by an ordinary solid-state sintering method, and their microstructures, dielectric, piezoelectric and ferroelectric properties were studied in detail. The effects of the substitutions on the phase transition were also investigated.

2. Experimental

The Sr_{2-x}Ca_xNaNb₅O₁₅ + y wt% MnO₂ (abbreviated as SCNNM-x/y) ceramics were prepared by the conventional ceramic fabrication technique. SrCO₃, Na₂CO₃, CaCO₃, Nb₂O₅, MnO₂ with the purity of over 99.9% were used as raw materials. The powders in stoichiometric ratio of Sr_{2-x}Ca_xNaNb₅O₁₅ were ball-milled for 24 h and calcined at 1150 °C for 12 h. After the calcination, MnO₂ was added. The resulting mixture was ball-milled again for 24 h and mixed thoroughly with a PVA binder solution, and then pressed into pillar with dimension 13 mm and thickness 20 mm under a uniaxial pressure of 100 Mpa. The compacted pillar samples were sintered at 1230–1300 °C for 3–7 h in air. The pillar samples were cut in the direction parallel to the pressing direction. The bulk samples were then machined, polished and coated golden paste on the large surfaces as electrodes. The ceramics were poled under a dc field of about 3 kV/mm at 200 °C in a silicone oil bath for 30 min.

The bulk densities ρ of samples were measured by the Archimedes method. The crystal structures of sintered samples were confirmed using an X-ray diffraction (XRD) meter with a nickel-filtered Cu-K α radiation (Bruker D8 Discover). The microscopy was observed using a scanning electron microscopy (SEM, JSM 6500 F). The relative permittivity ϵ_r and loss tangent $\tan \delta$ were measured as a function of temperature using an impedance analyzer (Agilent 4284A). A conventional Sawyer–Tower circuit was used to measure the polarization hysteresis (P – E) loop at 5 Hz. The piezoelectric coefficient d_{33} , the electromechanical coupling factors k_p and k_t were determined by the resonance and anti-resonance frequency method according to IEEE standard 176 using an impedance analyzer (Agilent 4294 A).

3. Results and discussions

The XRD patterns of the SCNNM-x/0.5 and SCNNM-0.05/y ceramics in the 2θ range of 20–60° are shown in Fig. 1. All the ceramics possess a pure tungsten–bronze structure, indicating that Ca²⁺ and Mn ions have diffused into the Sr₂NaNb₅O₁₅ lattices to form a new homogeneous solid solution. As the sintering temperature is high up to 1230 °C, Mn ions may exist in the SCNNM structure in two valence states: Mn³⁺ with a radius of 0.66 Å and Mn²⁺ with radius of 0.80 Å. Nb⁵⁺ has a radius of 0.69 Å which is close to 0.66 Å of Mn³⁺ and 0.80 Å of Mn²⁺, and smaller than Sr²⁺ (1.44 Å), Ca²⁺ (1.34 Å) and Na⁺ (0.97 Å). Therefore, according to the principles of crystal chemistry and radius matching ruler, Mn³⁺ or Mn²⁺ most likely enter into the A-sites for substituting Na⁺ and B-sites for substituting Nb⁵⁺, with Ca²⁺ substituting Sr²⁺. Fig. 1(b) shows the enlarged XRD patterns of the ceramics in the ranges of 2θ from 27° to 33°. At $x \geq 0.20$, the diffraction peaks (4 1 0) and (3 1 1) both split into two peaks, suggesting that the ceramics have a tendency to transform into another phase, e.g., orthorhombic, at high Ca²⁺ substitution levels [23]. It has been reported that the crystal structure of SCNN ceramics is tetragonal tungsten–bronze phase at room temperature, because it suppressed the tetragonal–orthorhombic phase transition and thus the tetragonal phase is maintained [17]. Together with these results, it suggests that the orthorhombic and tetragonal phases coexist in the SCNNM ceramics with $x \geq 0.20$ at room temperature. It can be noted that the diffraction peaks, especially, the (2 1 1) and (3 1 1) peaks shift slightly to higher 2θ as x increases. This may be attributed to the smaller ionic radius of Ca²⁺ as compared to Sr²⁺. It has been also observed that the XRD patterns of the ceramics with different y are similar (Fig. 1(c)). This indicates that the MnO₂ addition has no significantly effect on the crystal structure. Fig. 2 shows the scanning electron microscopy (SEM) micrographs of the SCNNM-x/y ceramics with $x/y = 0.05/0.5$, 0.25/0.5, 0.35/0.5, 0.05/0, 0.05/1.0, and 0.05/1.5. All the ceramics are dense and well crystallized, except for SCNNM-0.05/0 and -0.05/1.5. For the SCNNM-x/0.5 samples, there is an inhomogeneous grain size distribution, such as the grains have a bimodal grain size distribution, with big grains being surrounded by small grains. For the ceramic with $x = 0.25$ (Fig. 2(b)), as an example, the average

grain sizes are 12.2 and 5.1 μm for the big and small grains, respectively. It can be also seen that with the increasing of x (i.e., the concentration of Ca²⁺), the grains are nonuniform and the grain size remains almost unchanged, suggesting that the Ca²⁺ substitution has no significant effects on the densification of the ceramics. Unlike Ca²⁺, Mn ions have much more significant effects on the sintering of the SCNNM ceramics (Fig. 2(e–f)). Although the sintering temperature is high (1300 °C), the ceramic of SCNNM-0.05/0 has a relatively loose structure and a number of pores can be found (Fig. 2(d)). However, after the addition of 0.5 wt% MnO₂, the ceramics can be well sintered at a low sintering temperature of 1230 °C and a dense microstructure with enlarged grains is developed (Fig. 2(a)). Furthermore, the size of the grains is generally found obviously increased with the MnO₂ additive. One possible reason to trigger the grain growth is that the MnO₂ additive acting as the liquid phase during sintering above 1200 °C. The increasing grain size with the MnO₂ additive also implies that Mn ions were homogeneously dissolved into SCNNM grains within the solubility limit, while, the excess Mn ions accumulated at the grain boundaries to facilitate the liquid-phase sintering. However, when the amount of MnO₂ addition is over 1.0 wt%, some Mn ions may accumulate in the grain boundaries, and result in a pinning effect on the domain wall, hindering the motion of the domain [24]. As shown in Fig. 2(f), cracks are found in the SCNNM-0.05/1.5 ceramic and the grain size varies considerably with anomalous grain size distribution. This may be attributed to the reduction in the mobility of the grain boundary and the weakness in the mass transportation after the further addition of MnO₂. It is clearly seen that with a small amount of MnO₂ (0.3–1.0 wt%) doping, the SCNNM ceramics can be well sintered at a relatively low temperature. However, further addition of MnO₂ (over 1.0 wt%) would inhibit grain growth due to the accumulation of Mn ions at the grain boundaries, resulting a poor sintering performance of the ceramics.

The bulk density ρ , the axial ratio $\sqrt{10c/a}$, and the lattice unit volume V of sintered SCNNM ceramics are shown in Fig. 3. As shown in Fig. 3(a), the bulk density ρ for the SCNNM-x/0.5 ceramics is generally found decreases with increasing x (the concentration of Ca²⁺). On the other hand, the value axial ratio $\sqrt{10c/a}$ first increases slightly and decreases and then remains almost unchanged at $x \geq 0.20$. At the same time, the lattice unit volume V first increases and then decreases gradually with increasing x . The changes in the axial ratio $\sqrt{10c/a}$, and the lattice unit volume V should be related to a small difference in the ionic radii between Ca²⁺ (1.34 Å) and Sr²⁺ (1.44 Å). Moreover, in the tungsten–bronze system, the value of the axial ratio $\sqrt{10c/a}$ is usually used to evaluate the distortion of the NbO₆ octahedron. As Ca²⁺ entered Sr²⁺ site, the A-site disorder degrees and local compositional fluctuation may increase, thus giving rise to a normal phase transition. As shown in Fig. 3(b), it has been observed that the SCNNM-0.05/0 (i.e., SCNN) ceramic sintered at 1300 °C for 7 h has a low bulk density of 4.65 g/cm³. However, after the doping of 0.3–0.5 wt% MnO₂, the SCNNM ceramics give a high density over 4.86 g/cm³ even though sintered at a relatively low temperature of 1230 °C for 3 h. Furthermore, the axial ratio $\sqrt{10c/a}$, and the lattice unit volume V for SCNNM-0.05/y ceramics are generally found to change slightly with the increase of y (the doping lever of MnO₂). On the basis of these results and SEM observations, it is clearly seen that Ca²⁺ has no appreciable influence on the microstructure of the ceramics, while, the addition of MnO₂ effectively decreases the sintering temperature, produces a large-grained microstructure, and promotes the densification of the ceramics.

Fig. 4 shows the temperature dependence of the dielectric constant ϵ_r for the SCNNM-x/0.5 ceramics at 10 kHz as a function of temperature over the range of –200 to 350 °C. Similar to pure SCNN ceramic, the SCNNM-x/0.5 ceramics (i.e., SCNN ceramics with 0.5 wt% MnO₂ doping) undergo two phase transitions: one is asso-

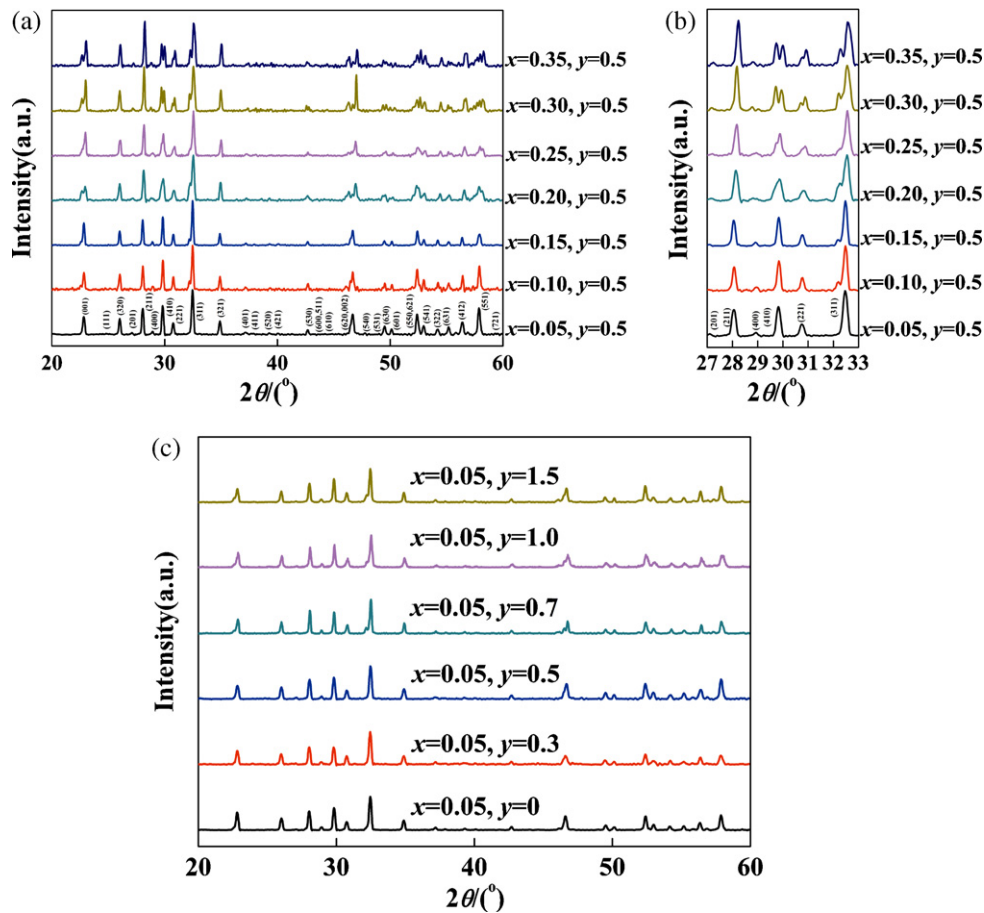


Fig. 1. X-ray diffraction (XRD) patterns of the SCNNM ceramics, and the enlarged XRD patterns indicating the shift of (2 1 1) and (3 1 1) peak.

ciated with the ferroelectric ($4mm$) to paraelectric ($4/mmm$) phase transition at the Curie temperature (T_c), and the other one is the ferroelectric ($mm2$) to ferroelectric ($4mm$) phase transition at lower temperature (T_s). The temperature corresponding to the maximum value of dielectric constant is named as the maximum temperature T_m . After the modifications with Ca^{2+} and MnO_2 , the SCNNM ceramics exhibit similar temperature dependence of ϵ_r , but with different T_c and T_s . It can be also noted that, T_c shifted to higher temperature and T_s shifted to lower temperature as x increasing from 0.05 to 0.35. Fig. 5(a) and (b) shows, as an example, the temperature dependence of ϵ_r for the SCNNM-0.10/0.5, 0.35/0.5 ceramics, while, the T_c and T_s with concentrations of Ca^{2+} (x) and MnO_2 (y) are summarized in Fig. 6.

As Fig. 5(a) and (b) shows, the ceramics present the main characteristic of relaxor ferroelectric, namely, strong frequency dependence of T_m , and different temperatures of maximum for ϵ_r and loss tangent $\tan \delta$ at a fixed frequency [25]. Moreover, as the substitution of Ca^{2+} increases, especially at $x \geq 0.20$ (as shown in Fig. 4), the ferroelectric–paraelectric phase transition became weak, and phase transition temperature T_c shifted to a temperature about 300°C , exhibiting a broaden and smooth transition peak in the plot of ϵ_r versus temperature. These provide additional evidence for the suggestion that the orthorhombic and tetragonal phases should coexist in the ceramics with $x \geq 0.20$ near room temperature. This is consistent with the results of X-ray diffraction (Fig. 1(a) and (b)).

The variations of T_c and T_s with x are summarized to form a phase diagram as shown in Fig. 6(a) for SCNNM- x /0.5 ceramics. As x increases from 0.05 to 0.20, T_c starts to increase from 260 to 290°C , then has a small decrease, and then increases with the increasing

of x , giving a maximum value of 299°C at $x = 0.35$. Unlike T_c , T_s first decreases, then increases and comes to a highest value of 8°C at $x = 0.20$, and then decreases rapidly with the addition of Ca^{2+} . These are attributed to a partial substitution of A-site ions Sr^{2+} by Ca^{2+} . For all the samples of SCNNM- x /0.5 ceramics except for $x = 0.20$, the observed T_s is below 0°C and at a range from -27 to -7°C , suggesting that the orthorhombic and tetragonal phase coexist in the ceramics near room temperature.

Fig. 5(c) and (d) shows temperature dependence of dielectric constant ϵ_r and the loss tangent $\tan \delta$ of SCNNM-0.05/0.3 and SCNNM-0.05/1.0 ceramics at different frequencies as a function of temperature over the range of -200 to 350°C , respectively. It can be seen that the phase transitions for the ceramic of SCNNM-0.05/0.3 and SCNNM-0.05/1.0 are similar, and the phase transitions temperature T_c and T_s changed only slightly. The variations of T_c and T_s with y for the SCNNM-0.05/ y ceramics are summarized to form a phase diagram as shown in Fig. 6(b). It can be seen that T_c and T_s both decrease rapidly as increasing y , and then increase, giving a minimum value of 237 and -26°C at $y = 0.10$, respectively. Compared with pure SCNN ceramic, the T_c and T_s of SCNNM ceramics both shift to lower temperature.

The diffuseness of a phase transition can be determined from the modified Curie–Weiss law $1/\epsilon - 1/\epsilon_m = C^{-1}(T - T_m)^\gamma$, where ϵ_m is the maximum value of dielectric constant at the phase transition temperature T_m , γ is the degree of diffuseness, and C is the Curie-like constant. The γ can have a value ranging from 1 for a normal ferroelectric to 2 for an ideal relaxor ferroelectric. Fig. 7 shows the plots of $\ln(1/\epsilon - 1/\epsilon_m)$ vs $\ln(T - T_m)$ for the SCNNM ceramics. As shown in Fig. 7(a), as x increases from 0.05 to 0.35, γ decreases obviously from 1.79 to 1.21 for the SCNNM- x /0.5 ceramics. This suggests

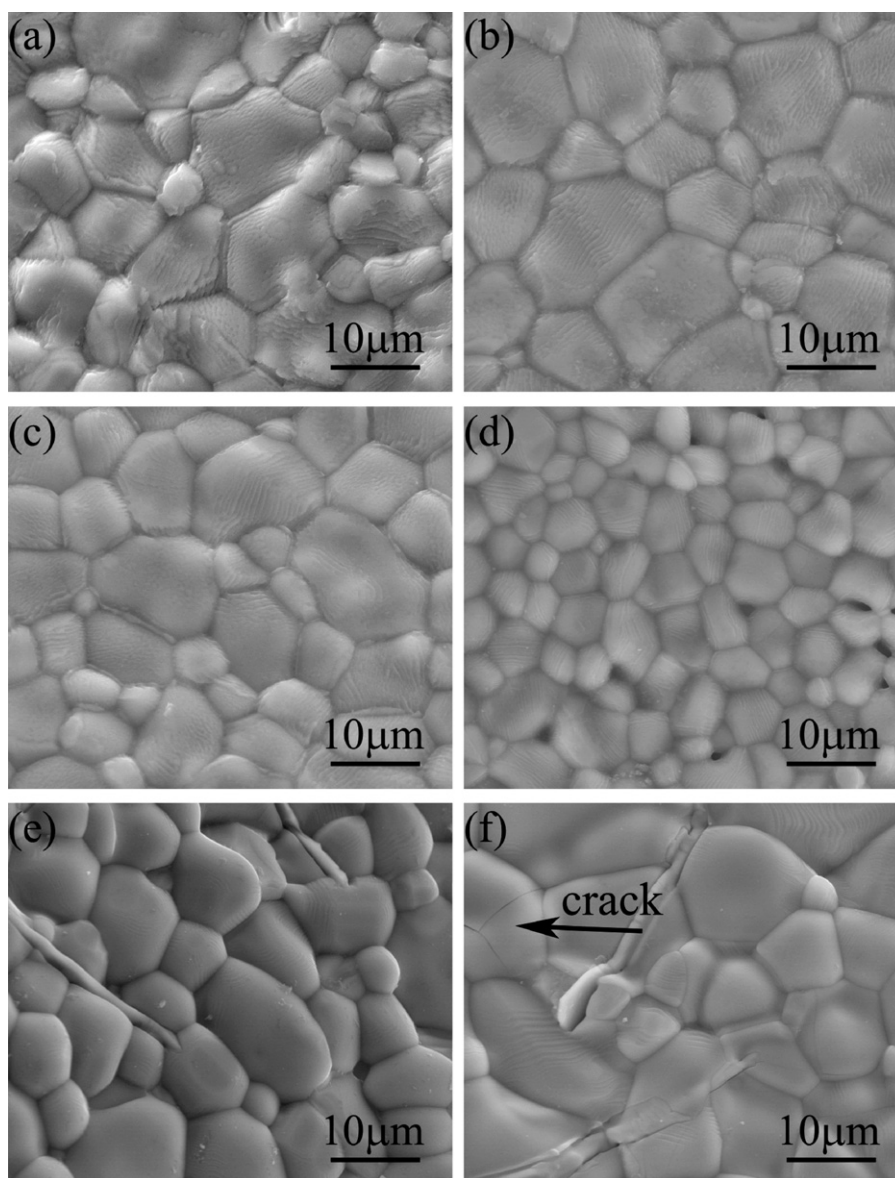


Fig. 2. SEM micrographs of the SCNNM- x/y ceramics: (a) $x=0.05, y=0.5$, sintered at $1230\text{ }^{\circ}\text{C}$ for 3 h; (b) $x=0.25, y=0.5$, sintered at $1230\text{ }^{\circ}\text{C}$ for 3 h; (c) $x=0.35, y=0.5$, sintered at $1230\text{ }^{\circ}\text{C}$ for 3 h; (d) $x=0.05, y=0$, sintered at $1300\text{ }^{\circ}\text{C}$ for 7 h; (e) $x=0.05, y=0.10$, sintered at $1230\text{ }^{\circ}\text{C}$ for 3 h; (f) $x=0.05, y=0.15$, sintered $1230\text{ }^{\circ}\text{C}$ for 3 h.

that the substitution of Ca^{2+} makes the ceramic have transformed gradually from a relaxor ferroelectric to a normal ferroelectric, thus exhibiting a weakened and broadened transition peak in the plot of ε_r versus temperature, as shown in Fig. 5(a) and (b). For the SCNNM- $x/0.5$ ceramics, Ca^{2+} (1.34 \AA) substitute the A-site ions Sr^{2+} (1.44 \AA). As the radius difference between the cations increases with increasing of smaller Ca^{2+} substitution for larger Sr^{2+} , the octahedral tilting likely becomes easier, which formed more stable polar clusters, then the coupling between polar clusters becomes stronger and makes the ceramics change from a diffused feature into a normal ferroelectric phase transition [26,27]. The values of γ for all the SCNNM-0.05/ y ceramics are higher than 1.79, and changed slightly with y increasing, indicating that the MnO_2 substitution does not induce significant changes in the phase transition, and the SCNNM-0.05/ y ceramics are close to ideal relaxor ferroelectric.

The variation of the piezoelectric coefficient d_{33} , electromechanical coupling factors k_p and k_t , dielectric constant ε_r and loss tangent $\tan \delta$ with x for the SCNNM- $x/0.5$ ceramics are shown in Fig. 8. Poled specimens with $x=0.30$ and 0.35 give no piezoelectric response. As shown in Fig. 8(a), the SCNNM-0.05/ y ceramic exhibits similar com-

positional dependences of d_{33} , k_p , k_t . All of them decrease gradually with the increasing of x . On another hand, the relative dielectric constant ε_r (1 MHz) slightly increases with the increasing of x , and then decreases gradually at $x > 0.15$, giving a maximum value of 2226 at $x=0.10$. While, the loss tangent $\tan \delta$ (1 MHz) increases slightly from 0.038 to 0.051, with x increasing from 0.05 to 0.15, and then starts to decrease at $x=0.20$ (Fig. 8(b)). It should be noted that the ceramics with $x=0.05$ exhibit excellent characteristics such as d_{33} (~ 190), k_p ($\sim 13.4\%$), k_t ($\sim 36.5\%$), and high ε_r (~ 2123) and low loss tangent $\tan \delta$ (~ 0.038).

Fig. 9 shows the variations of d_{33} , k_p , k_t , ε_r , and the loss tangent $\tan \delta$ with y for the SCNNM-0.05/ y ceramics. It can be noted that the observed d_{33} , k_p , k_t , and ε_r exhibit similar dependences on y , i.e., first increases steeply with y increasing from 0 to 0.5, and then decreases, they all reach an optimum value, being 190 pC/N, 13.4%, 36.5%, 2123 at $y=0.5$, respectively. At the same time, the loss tangent $\tan \delta$ remains at a relatively low value (≤ 0.055) in the range of y from 0 to 1.5. However, without MnO_2 doping, the SCNNM-0.05/0 ceramic has a poor density and weak piezoelectricity ($\rho=4.53\text{ g/cm}^3$, $d_{33}=51\text{ pC/N}$, $k_p=6.9\%$, $k_t=19.2\%$). After

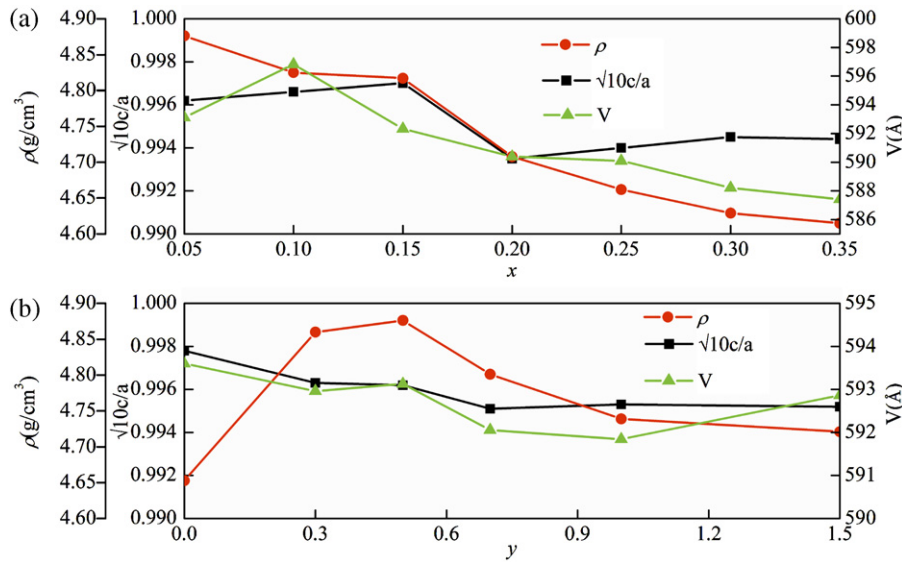


Fig. 3. Variations of the density ρ , the axial ratio $\sqrt{10c/a}$, and the volume for (a) SCNNM- $x/0.5$ ceramics and (b) SCNNM-0.05/ y ceramics.

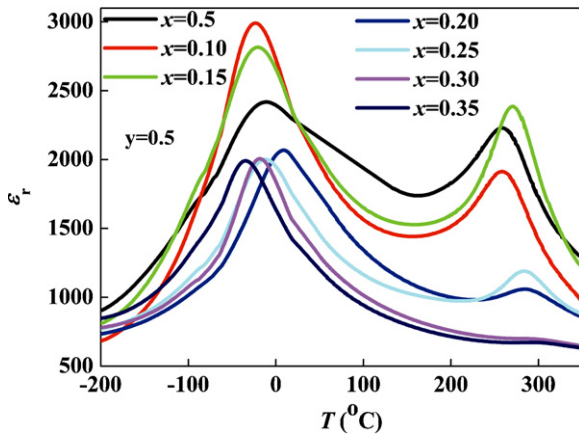


Fig. 4. Dielectric constant ϵ_r of the SCNNM- $x/0.5$ ceramics as a function of temperature at 10 kHz.

the use of MnO_2 additive (0.3–0.5 wt%), the density, the dielectric and piezoelectric properties are considerably enhanced. These indicate that MnO_2 is an effective sintering aid for improving the densification (Fig. 2(b)) and plays an important role in improvement of piezoelectric and dielectric properties (Fig. 8). Moreover, the increase in ϵ_r can be understood by the increase in the grain size with MnO_2 addition. The MnO_2 additive enhances the average grain size, making the domain wall motion easier, resulting in an increased dielectric coefficient. Meanwhile, the loss tangent $\tan \delta$ decreases from 0.055 to 0.028 with y increasing from 0 to 0.10, and then increases slightly from 0.028 to 0.034 at $y \geq 0.10$. For a small amount of MnO_2 additive (0.3–1.0 wt%), the systems show simultaneously “softening” and “hardening”, which resulting in the variations of piezoelectric properties [28]. The “soft” and “hard” dopant mechanisms of MnO_2 in the ceramics will be shown in the next section. Further addition of MnO_2 above 1.0 wt% led to the decrease of piezoelectricity because of nonuniformity of the microstructure.

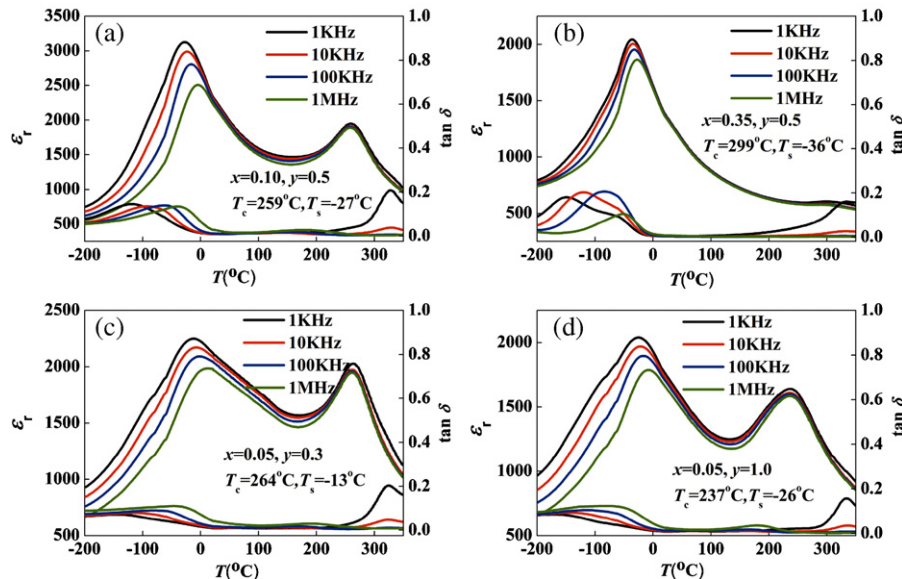


Fig. 5. Dielectric constant ϵ_r and loss tangent $\tan \delta$ of SCNNM ceramics as a function of temperature: (a) $x=0.10$, $y=0.5$; (b) $x=0.35$, $y=0.5$; (c) $x=0.05$, $y=0.3$; (d) $x=0.05$, $y=1.0$.

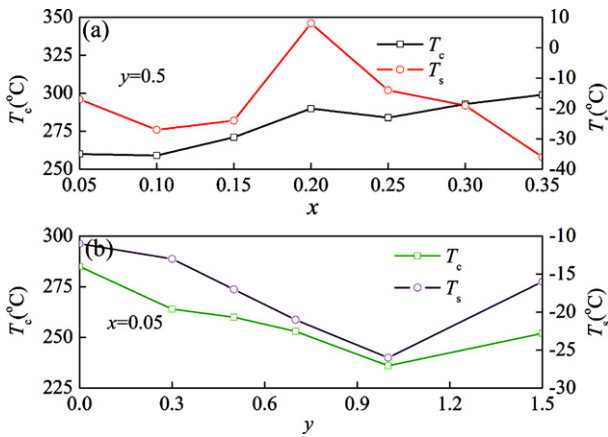


Fig. 6. Variations of T_c and T_s of (a) SCNNM- x /0.5 ceramics, with $x=0.05$ –0.35, and of (b) SCNNM-0.05/ y ceramics, with $y=0$ –1.5.

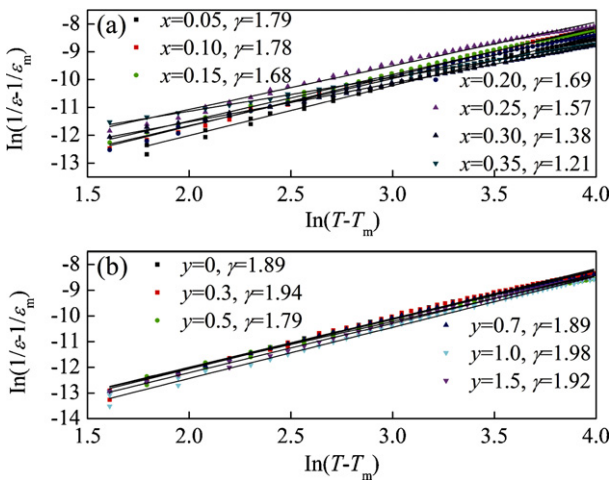


Fig. 7. Plot of $\ln(1/\epsilon - 1/\epsilon_m)$ vs $\ln(T - T_m)$ for the SCNNM- x / y ceramics. The symbols denote experiment data, while the solid lines denote the least-squares fitting line to the modified Curie–Weiss law.

Figs. 10(a) and 11(a) show the typical P - E hysteresis loops of SCNNM- x /0.5 and SCNNM-0.05/ y ceramics, measured under an electric field of about 3 kV/mm at room temperature. The variations of the remanent polarization P_r and coercive field E_c with x and y are shown in Figs. 10(b) and 11(b), respectively. It is clearly

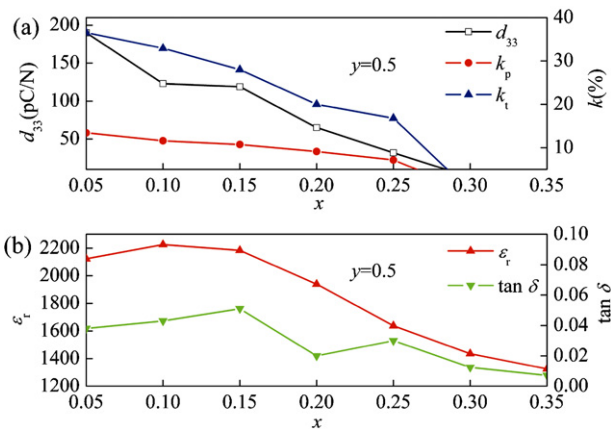


Fig. 8. (a) Variations of piezoelectric coefficient d_{33} and electromechanical coupling factors k_p and k_t with x for the SCNNM- x /0.5 ceramics; (b) variations of the dielectric constant ϵ_r and loss tangent $\tan \delta$ with x for the SCNNM- x /0.5 ceramics.

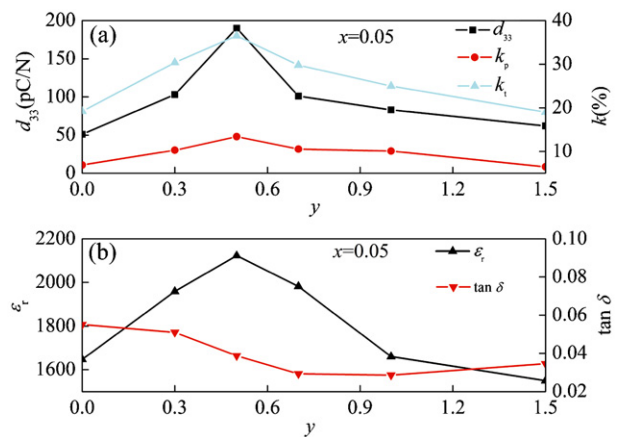


Fig. 9. (a) Variations of piezoelectric coefficient d_{33} and electromechanical coupling factors k_p and k_t with y for the SCNNM-0.05/ y ceramics; (b) variations of the dielectric constant ϵ_r and the loss tangent $\tan \delta$ with y for the SCNNM-0.05/ y ceramics.

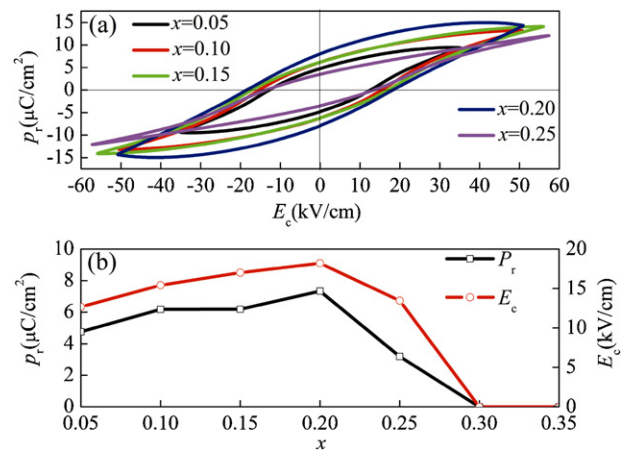


Fig. 10. (a) P - E hysteresis loops of the SCNNM- x /0.5 ceramics with different x ; (b) variations of the remanent polarization P_r and coercive field E_c with x for the SCNNM- x /0.5 ceramics.

to see that for the SCNNM- x /0.5 ceramics at $x=0.05$ –0.25, the P - E loops are slim, slanted, and well saturated. Similar to the piezoelectric properties, specimens with composition of $x=0.30$ and 0.35 give no ferroelectric properties. For the SCNNM- x /0.5 ceramics, the

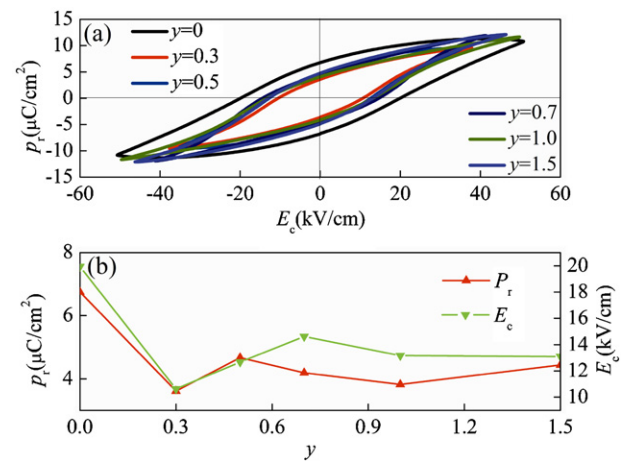


Fig. 11. (a) P - E hysteresis loops of the SCNNM-0.05/ y ceramics with different y ; (b) variations of the remanent polarization P_r and coercive field E_c with y for the SCNNM-0.05/ y ceramics.

observed E_c increases gradually with x increasing from 0.05 to 0.20, and then starts to decrease rapidly, giving a maximum value of 18.22 kV/cm at $x=0.20$. The remnant polarization P_r shows similar dependence on x , i.e., first increases from 4.76 to 7.34 $\mu\text{C}/\text{cm}^2$ as x increases from 0.05 to 0.20, and then decreases linearly at $x \geq 0.20$. These clearly show that the substitution of Ca^{2+} has enhanced the ferroelectric properties first, and then weakened the ferroelectric properties of the SCNNM ceramics. The enhancement of remnant polarization P_r with Ca^{2+} additive may be related to the enhanced domain wall mobility, which allows for more complete domain switching of the ceramics. Similarly, all the SCNNM-0.05/ y ceramics exhibit well-saturated P - E loops under an electric field of about 3 kV/mm. Fig. 11(b) shows the remnant polarization P_r and coercive field E_c of the SCNNM-0.05/ y ceramics as a function of y . It can be seen that P_r decreases from 6.75 to 3.14 $\mu\text{C}/\text{cm}^2$, as y increases from 0 to 0.3, and then increases slightly and becomes saturated at about 3.80 $\mu\text{C}/\text{cm}^2$. E_c is lowered greatly from 19.94 to 10.63 kV/cm, as y increases from 0 to 0.3, and then increases and decreases slightly. Compared with the pure SCNN ceramics, the SCNNM ceramics have lower coercive field E_c and relatively good ferroelectric properties. The observed dielectric, piezoelectric and ferroelectric properties show that Mn ions can cause both “soft” and “hard” doping effects simultaneously in SCNNM-0.05/ y ceramics. In a sense, at low MnO_2 additive ration (≤ 0.5 wt%), Mn ions go to Na^+ sites in SCNNM. It is well known that Na_2O is volatile at high temperature. Na^+ in SCNNM may leave the ceramic and form some vacancies in the lattice during the sintering process. The substitution of Na^+ (0.97 Å) by higher valence of Mn^{3+} (0.66 Å) and Mn^{2+} (0.80 Å), creates cation vacancies, thus, Mn ions act as “donors”. Ions in the lattice are transferred easier due to the presence of vacancies, which results in the easy movement of domain walls, which makes the ferroelectric ceramic to become “softer”. When the MnO_2 additive ratio between 0.5 and 1.0 wt%, B-site ions Nb^{5+} (0.69 Å) may be substituted by Mn^{3+} (0.66 Å), thus causes a fast drop in d_{33} , k_t , ε_r and a slight decrease in P_r , indicating that MnO_2 additives have played as a hard agent in the SCNNM ceramics. In conclusion, Mn doping in SCNNM ceramics produced both soft and hard effects, caused by the multivalence nature of Mn ions, which get into different sites in the tungsten-bronze structure.

4. Conclusions

Lead-free SCNNM- x/y piezoelectric ceramics have been prepared by an ordinary sintering technique, and their microstructures, phase transitions, and electrical properties have been investigated in detail. Our results reveal that Ca^{2+} and Mn ions have diffused into $\text{Sr}_2\text{NaNb}_5\text{O}_{15}$ lattices to form a solid solution with

tungsten-bronze structure. The substitution of Ca^{2+} decreases d_{33} , k_p and k_t , and makes the ceramics become less relaxor. A small amount of MnO_2 (0.3–1.0 wt%) effectively promotes the densification and decreases the sintering temperature of the ceramics. The densification, “hard” and “soft” effects of MnO_2 doping should be attributed to the strong dielectricity and high piezoelectricity of the ceramics. For the ceramics with $x=0.05$, $y=0.5$, the piezoelectric and dielectric properties become optimum, giving $d_{33} = 190$ pC/N, $k_p = 13.4\%$, $k_t = 36.5\%$, $\varepsilon_r = 2123$, loss tangent $\tan \delta = 0.038$ and the Curie temperature $T_c = 260$ °C. Results show that MnO_2 doped $\text{Sr}_{2-x}\text{Ca}_x\text{NaNb}_5\text{O}_{15}$ ($x=0.05$ –0.35) ceramics are one of the good candidates for lead-free piezoelectric material.

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